MP BOARD CLASS 10 EM SCIENCE MODEL PAPER SET 4 2020

म.प्र. बोर्ड कक्षा 10 EM विज्ञान मोडल पेपर सेट 4 2020

Time: 3 Hours) (Max. Marks: 80 Instructions: (1) All questions are compulsory. (2) There are four types of objective questions from Q. No. 1 to 4. (3) Internal choices are given in question No. 5 to 22. (4) Marks of each question are indicated against it. (5) Draw neat and labelled diagram where necessary. 1. Choose and write the correct alternative : 5x 1 = 5(i) $Fe_2O_3 + 2A1 \rightarrow A1_2O_3 + 2Fe$ The above reaction is an example of a: (a) combination reaction. (b) double displacement reaction (c) decomposition reaction (d) displacement reaction. (ii) Which among the following is not a metalloid? (a) Silicon (b) Germanium (c) Boron (d) Sodium. (iii) Which of the following metals is associated with haemoglobin? (a) Aluminium (b) Potassium (c) Calcium (d) Iron. (iv) The brain is responsible for : http://www.mpboardonline.com (a) thinking (b) regulating heart beat (c) balancing the body (d) all of these. (v) The chromosome responsible for sex determination in a male is : (a) X (b) Y (c) XY (d) XX. Ans. (i) (d), (ii) (d), (iii) (d), (iv) (d), (v) (c). 2. Fill in the blanks : $5 \times 1 = 5$ (i) can dissolve gold and platinum. (ii) objects generate their own light. (iii) The colour is least scattered by fog or smog. (iv) Polythene is an example of waste. (v) are biodiversity hot spots. Ans. (i) Aqua regia, (ii) Luminous, (iii) Red, (iv) Non-biodegradable, (v) Forests. 3. Match the columns : $5 \times 1 = 5$ Column 'A' Column 'B' (a) V = IR(i) Fire extinguisher (b) Second tropic level (ii) Respiration (iii) Concave lens (c) Oxidation (iv) Ohm's law (d) Baking soda (v) Herbivores (e) Facilitates distant vision Ans. (i) \rightarrow (d), (ii) \rightarrow (c), (iii) \rightarrow (e), (iv) \rightarrow (a), (v) \rightarrow (b). 4. Answer in one word/sentence : $5 \times 1 = 5$

(i) What is the chemical formula of washing soda?

(ii) In which form glucose is stored for longer duration in plant cells?

(iii) What will be the 23rd chromosome pair for a female human body? (iv) What will happen if the intake of iodine in the diet is low? (v) What is a galvanometer? http://www.mpboardonline.com Ans. (i) Na₂CO₃, 10H₂O, (ii) As starch, (iii) XX, (iv) Its deficiency causes goitre, (v) An instrument used to detect the presence and direction of current in a circuit is called galvanometer. 5. Why does the colour of copper sulphate solution change when an iron nail is dipped in it? 2 Or Why do we apply paint on iron articles? 6. An atom has electronic configuration 2, 8, 7. What is the atomic number of this element? 2 Or Name -(a) three elements that have a single electron in their outermost shells. (b) two elements that have two electrons in their outermost shells. 7. How does binary fission differ from multiple fission? 2 What is the role of seminal vesicles and the prostate gland? 8. What are analogous organs? 2 Or Why are the traits acquired during the lifetime of an individual not inherited? 9. Define the principal focus of a concave mirror. 2 Or The magnification produced by a plane mirror is + 1. What does this mean? 10. Why should a magnesium ribbon be cleaned before it is burnt in air? 3 Or Explain the following terms of gain or loss of oxygen with two examples each: (a) Oxidation, (b) Reduction. 11. The far point of a myopic person is 80 cm in front of the eye. What is the nature and power of the lens required to correct the problem? Or Explain advanced sunrise and delayed sunset. 12. (a) Why does a compass needle get deflected when brought near a bar magnet? 3 (b) Why don't two magnetic lines of force intersect each other? Or State Fleming's left hand rule. 13. List the properties of magnetic lines of force. 3 Or Imagine that you are sitting in a chamber with your back to one wall. An electron beam, moving horizontally from back wall towards the front wall is deflected by a strong magnetic field to your right side. What is the direction of magnetic field? http://www.mpboardonline.com 14. If you could use any source of energy for heating your food, which one would you use and

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why?

Or

What are the advantages and disadvantages of using solar cooker? Are there places where solar cooker would have limited utility?

15. Why do HCI, HNO3 etc. show acidic characters in aqueous solutions, while solutions of compounds like alcohol and glucose do not show acidic character?

Or

Differentiate between acids and bases. Name a strong acid and a base.

16. An aldehyde as well as a ketone can be represented by the same molecular formula, say C_3H_6O . Write their structures and name them. State the relation between them in the language of science. 4 Or

How can ethanol and ethanoic acid be differentiated on the basis of their physical and chemical properties?

17. What are the differences between autotrophic nutrition and heterotrophic nutrition?

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Compare the functioning of alveoli in the lungs and nephrons in the kidneys with respect to their structure and functioning.

18. Write the main functions of cytokinins.

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Or

What is the function of receptors in our body? Think of situations where receptors do not work properly. What problems are likely to arise?

19. An electric motor takes 5 A from a 220 V line. Determine the power of the motor and the energy consumed in 2 hours.

Or

An electric heater of resistance 8 $\,\Omega$ draws 15 A from the series mains for 2 hours. Calculate the rate at which heat is developed in the heater.

20. Define the following terms:

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(i) Mineral, (ii) Ore, (iii) Gangue.

Or

- (i) Write the electron-dot structures for sodium, oxygen and magnesium.
- (ii) Show the formation of Na₂O and MgO by the transfer of electrons.
- (iii) What are the ions present in these compounds?
- 21. Explain the term regeneration as used in relation to reproduction of organisms. Descibe briefly how regeneration is carried out in multicellular organisms like Hydra.

Or

List six specific characteristics of sexual reproduction. http://www.mpboardonline.com

22. The absolute refractive indices of glass and water are 4/3 and 3/2 respectively.

If the speed of light in glass is 2 x 10⁸ m/s. Calculate the speed of light in (i) vacuum, (ii) water.

Or

A 4 cm tall object is placed perpendicular to the principal axis of a convex lens of focal length 20 cm. The distance of the object from the lens is 15 cm. Find the nature, position and size of the image.