CHEMISTRY - 2013

| Time: 3 | Hours | Class | : 12th | M. M.: 75 | | |
|---------|---|---|------------------------|-------------------|--|--|
| Note- | (i) | All questions are con | ipulsory. | | | |
| | (ii) | Read the instructions their answers. | s of question paper ca | refully and write | | |
| | (iii) | There are two Section question paper. | ons- Section - A and | Section-B in the | | |
| | (iv) | In Section-A Q. Nos. 1 to 4 are objective type questions which contains choose the correct answers, match the columns, fill up the blanks and one word answer. Each question carries 5 marks. | | | | |
| | (v) | Internal choices are given in Q. Nos. 5 to 17 of Section-B. | | | | |
| | (vi) | Q. Nos. 5 to 14 carry 4 marks each. | | | | |
| | (vii) | Q. Nos. 15 to 17 carry | y 5 marks each. | | | |
| | | (Sectio | n - A) | | | |
| | | (Objective Ty | pe Questions) | | | |
| Q. 1. | Write the correct answer from the given options which provided in | | | | | |
| | every objective type question: | | | | | |
| | (A) | Which of the following is ferromagnetic? | | | | |
| | | (i) Calcium metal | (ii) Iron metal | | | |
| | | (iii) Sodium metal | (iv) Zinc metal | | | |
| | (B) | Silicon is: | | | | |
| | | (i) Good conductor | (ii) Bad conduc | ctor | | |
| | | (iii) Semiconductor | (iv) Ore | | | |
| | (C) | The crystal lattice of NaCl is: | | | | |
| | | (i) Face-centered cubic lattice (ii) Body-centered cubic lattice | | | | |
| | | | | | | |
| | | (iii) Simple cubic lattice | | | | |
| | | (iv) Hexagonal close packing | | | | |
| | (D) | crystal? | | | | |
| | | (i) Frenkel | (ii) Schottky | - | | |
| | | (iii) Linear | (iv) Impurity | | | |
| | (E) | Brass is: | • | | | |
| | | (i) Solid solution | (ii) Liquid solu | tion | | |
| | | (iii) Gas solution | (iv) All of these | е | | |

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| Q. 2. | Match the pairs correctly (choose the correct answer from Section 'B' for Section 'A'): | | | | |
|-------|---|--------------------------------------|--|--|--|
| | Section 'A' | Section 'B' | | | |
| - | (A) One Faraday | (i) H ₂ SO ₄ . | | | |
| | (B) Arrhenius equation | (ii) HF | | | |
| | (C) Haematite | (iii) 96500 coulombs | | | |
| | (D) Contact process | (iv) $K = Ae^{-Ea/RT}$ | | | |
| | (E) Noble gas | (v) Iron | | | |
| • | | (vi) $H_2S_2O_8$ | | | |
| | | (vii) Group 18 | | | |
| Q. 3. | Fill in the blanks: | | | | |
| | (A) 'F' block elements are known as elements. | | | | |
| | (B) The basic characte | r of Amine is due to the presence of | | | |
| | en atom. | | | | |
| | (C) Starch is saccharde | | | | |
| | (D) The substance which low e acidity of stomach is | | | | |
| | called | • | | | |
| | (E) Catalysts which incre | ase rate of a reaction are called | | | |
| | catalyst. | | | | |
| Q.4. | Q.4. Write answers in one sentence each: | | | | |
| | (i) Who used the word 'catalyst' for the first time? | | | | |
| | (ii) Why does colloidal particle show Tyndall effect? | | | | |
| | (iii) Cleansing action of soap based on which principle? | | | | |
| | (iv) Why does aniline turn blackish brown in open air? | | | | |
| | (v) What is the main pro | oduct of mustard oil reaction? | | | |
| | (Sec | tion-B) | | | |
| | (Very Short Answ | ver Type Questions) | | | |
| Q. 5. | Write short notes on: | | | | |
| | (i) Activation energy | | | | |
| • | (ii) Arrhenius equation. | (Or) | | | |
| | A first order reaction is completed 90% in 40 minutes. Calculate its | | | | |
| | half-life period. (log 10 = 1) | | | | |
| Q. 6. | Explain Siemens- Martin's open hearth process for the manufacture | | | | |
| - | of steel with diagram. (Or) | | | | |
| | What is Blue Vitriol? Write effect of head on it. | | | | |
| | | • | | | |

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(Or) .

| Q. 7. | Describe Brodie's ozonizer with diagram (Bleaching of flowers by Cl, is permanent while bleaching by SC | Or) | | | |
|--------|--|------|--|--|--|
| | temporary. Why? | | | | |
| Q. 8. | Xenon is a noble gas but it forms compounds, why? Draw the struc- | | | | |
| | ture of any two compounds of it. (Or) | | | | |
| | Explain the following: | | | | |
| | (i) Available Chlorine | | | | |
| | (ii) Causes of inertness of noble gases. | | | | |
| Q. 9. | What are chelates? Giving one example write the importance of chelate. | | | | |
| | | Or) | | | |
| | What is Zeise's Salt and Ferrocene? Explain with structure. | | | | |
| Q. 10. | Write the chemical reaction of chloroform with the following: | | | | |
| * | (i) Silver powder | | | | |
| | (ii) Concentrate HNO ₃ | | | | |
| | (iii) Acetone | | | | |
| * | (iv) Alkaline Solution of Phenol. | Or) | | | |
| | Why are Haloarenes less reactive than Haloalkane? | | | | |
| Q. 11. | What is Lucas reagent? Distinguish primary, secondary and tertiary | | | | |
| | alcohol using Lucas reagent. | Or) | | | |
| - | Ethyl alcohol and Phenol both contain OH group. Why Pheno | l is | | | |
| | acidic and alcohol is neutral in nature? Give reason. | | | | |
| Q. 12. | How will you obtain: | | | | |
| | (i) Formic acid from Acetic acid. | | | | |
| | (ii) Acetaldehyde from Acetylene. | Or) | | | |
| | Explain the preparation of Acetic acid by quick vinegar process w | vith | | | |
| | diagram. http://www.mpboardonline.com | | | | |
| Q.·13. | What is denaturation of protein and name the two diseases cause | sed | | | |
| | | Or) | | | |
| | What are enzymes? Give two examples with uses. | | | | |
| Q. 14. | Write notes on: | | | | |
| | (i) Kanad (ii) Sushrut. | Or) | | | |
| • | What is preservative? Give the name and formula of any two preservatives. | ore- | | | |
| | (Short Answer Type Questions) | | | | |
| Q. 15. | What is Raoult's law? How can molar mass of a non-volatile solute | e be | | | |

determined with its help?

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Calculate the osmotic pressure of 5% solution of glucose at 25° C (Molecular weight of Glucose = 180 and R = 0.0821 liter atm K⁻¹ mol⁻¹)

Q. 16. Explain the dry cell with diagram.

(Or)

Write short notes on:

- (i) Nernst equation
- (ii) Corrosion.
- Q. 17. Explain the following:

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- (i) Cu⁺ is colourless and Cu²⁺ is coloured
- (ii) Zn shows only +2 oxidation state. (Or)

Compare Lanthanide and Actinide. (Write any five points).

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